

2,5-S,S-DICYSTEINYLDOPA: A NEW AMINO ACID IN THE EYE OF THE GAR
AND ITS ENZYMIC SYNTHESIS

Shosuke Ito and J. A. C. Nicol

The University of Texas Marine Science Institute, Port Aransas,
Texas 78373, U.S.A.

(Received in USA 23 July 1975; received in UK for publication 12 August 1975)

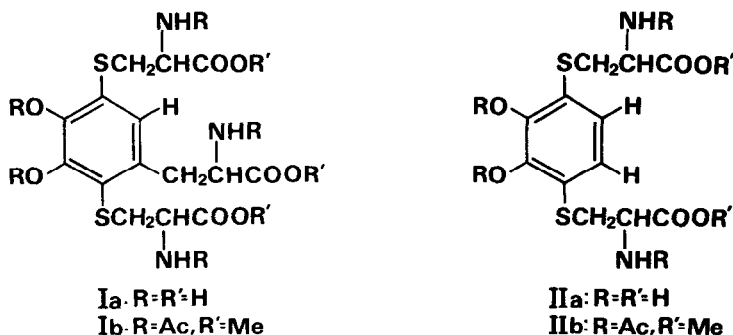
In the course of a study of reflecting materials in fish eyes¹, we have isolated from the alligator gar Lepisosteus spatula a new amino acid whose structure has been established as 2,5-S,S-dicysteinyldopa (Ia).

An acid extract (1M HCl) of 50 eyes was passed on to a Dowex-50W x4 (H⁺ form) column and, after washing with 1M HCl, the amino acid was eluted with 6M HCl, and then chromatographed on Sephadex LH-20, using MeOH:1M HCl 19:1 as eluent. The amino acid was further purified by paper chromatography in n-PrOH:1M HCl 3:2 and converted to the formate by passing through a Dowex-2 x8 (formate form) column. Chromatography on Sephadex G-25 in 0.5M HCOOH yielded the amino acid (26 mg) as an amorphous colorless powder. It gave a violet color with ninhydrin and a pale green one with FeCl₃ and was pure as judged by t.l.c. on cellulose. The u.v. maxima are given in the Table.

The elemental analysis showed the molecular formula to be C₁₅H₂₁N₃O₈S₂·HCOOH (Found C, 39.67; H, 5.27; N, 8.76; S, 13.33%; calc.: C, 39.90; H, 4.81; N, 8.73; S, 13.32%). The n.m.r. spectrum (100 MHz, 1M DCl) revealed, besides the formate proton, only ten protons; six methylene protons (δ 3.2-3.8, m), three methine protons (δ 4.3-4.5, m), and one isolated aromatic proton (δ 7.20, s). These data suggested that the amino acid Ia might be derived from one molecule of dopa and two molecules of cysteine. Indeed, reduction with 5% HI-red P² (110°, 24 hr) yielded three amino acids, cysteine, dopa (ca. 50%) and 5-S-cysteinyldopa³ (ca. 20%).

In order to establish the structure and secure much more material, an attempt was made to synthesize the amino acid by enzymic oxidation. Under a carefully controlled

condition (22°, 10 hr), the reaction of L-dopa with L-cysteine (3 eq.) at pH 6.8 in the presence of mushroom tyrosinase and atmospheric O₂ yielded, along with 5-S-cysteinyl-dopa³ (68%) and 2-S-cysteinyl-dopa⁴ (5.8%), an amino acid (8.5%), C₁₅H₂₁N₃O₈S₂·HCOOH⁵, identical with the gar amino acid in spectral properties (u.v.(Table), i.r., and n.m.r.) and t.l.c. Esterification of the synthetic Ia with MeOH-HCl followed by acetylation with Ac₂O-pyr gave the N, O-pentacetyl trimethyl ester Ib (22%). The n.m.r. spectrum (CDCl₃) was consistent with the structure Ib; it revealed the presence of three acetamide groups (δ 1.82, 1.92, 2.04), two phenyl acetate groups (δ 2.37, 2.38), three carbomethoxy groups (δ 3.64, 3.74, 3.78), and one aromatic proton (δ 7.34, s). These results indicated that the amino acid might be 2,5-S,S- or 5,6-S,S-dicysteinyl-dopa.



The formation of 2,5-S,S-dicysteinyl-dopa can be explained by 1,6-addition³ of the second cysteine molecule to the conjugated o-quinone formed in situ from 5-S-cysteinyl-dopa, whereas that of 5,6-S,S-dicysteinyl-dopa can be explained by 1,4-addition of cysteine. That the former is a more likely case was proved by the enzymic synthesis of 3,6-S,S-dicysteinylcatechol, a product of two 1,6-additions. Reaction of catechol with L-cysteine (3 eq.) in the presence of tyrosinase and O₂ gave 3-S-cysteinylcatechol (41%), C₉H₁₁NO₄S·HCl⁵, m.p. 172-175° (dec.), and 3,6-S,S-dicysteinylcatechol (IIa) (17%), C₁₂H₁₆O₆S₂·2H₂O⁵, m.p. 262° (dec.). The structures of these catechol derivatives were confirmed by their spectral data, especially by n.m.r. spectra.

The presence of three adjacent protons in 3-S-cysteinylcatechol was shown by the n.m.r. spectrum (1M DCl): δ 6.86, dd, J = 8.5, 7.5 Hz; 7.01, dd, J = 8.5, 2 Hz, 7.09, dd, J = 7.5, 2 Hz. The n.m.r. spectrum (1M DCl) of 3,6-S,S-dicysteinylcatechol (IIa) indicated that it has a symmetrical structure: two equivalent methylene groups (δ 3.54, d, J = 5.6 Hz) two equivalent methine groups (δ 4.32, t, J = 5.6 Hz) and two equivalent aromatic protons

(δ 7.10, s). Esterification of IIa followed by acetylation yielded the O, N-tetracetyl dimethyl ester IIB (81%); M^+ , m/e 544. The symmetrical structure IIB was supported by the n.m.r. spectrum ($CDCl_3$): δ 1.76, 6H, s (NCOMe); 2.40, 6H, s (OCOMe); 3.37, 3.55, both 2H, both dd, $J = 14$, 4 Hz (CH_2); 3.76, 6H, s (COOMe); 4.88, 2H, dt, $J = 7.5$, 4 Hz (CH); 6.49, 2H, d, $J = 7.5$ Hz (NH); and 7.38, 2H, s (aromatic protons).

Table. The u.v. spectral data of the compounds Ia, Ib, IIa, IIB

compound	solvent	λ_{max}/nm (ϵ)
2,5-S,S-dicysteinyldopa (Ia)	0.1M HCl	217 (23400), 273 (8200), 303 (3100)
(natural)	pH 6.8 buffer	223 (22800), 267 sh. (7200), 316 (3300)
2,5-S,S-dicysteinyldopa (Ia)	0.1M HCl	217 (22700), 273 (8600), 302 (3100)
(synthetic)	pH 6.8 buffer	223 (22100), 267 sh. (6900), 316 (3200)
3,6-S,S-dicysteinyldopa (IIa)	0.1M HCl	212 (26600), 270 (10800), 298 sh. (3200)
dopa derivative Ib	MeOH	279 (10700)
catechol derivative IIB	MeOH	277 (13200)

Thus, the structure Ia, 3-(2,5-S,S-dicysteinyldopa-3,4-dihydroxyphenyl)-alanine, can be assigned to the new amino acid from the eye of the gar. The compound may be formed in vivo by two consecutive 1,6-additions of two cysteine molecules first to dopaquinone and then to 5-S-cysteinyldopaquinone produced by the action of tyrosinase. 5-S-Cysteinyldopa and 2-S-cysteinyldopa have been shown to be intermediates in the biosynthesis of phaeomelanine⁶.

We are grateful to Prof. G. Prota (Univ. of Naples, Italy) for helpful discussions. This work was supported by a grant from the National Eye Institute, National Institutes of Health, No. EY00495.

References and Note

- (a) S. Ito and J. A. C. Nicol, Biochem. J. **143**, 207 (1974).
(b) S. Ito and J. A. C. Nicol, Proc. Roy. Soc. Ser. B, **190**, 33 (1975).
- R. A. Nicolaus, G. Prota, C. Santacrose, G. Scherillo, and D. Sica, Gazz. Chim. Ital. **99**, 323 (1969).

3. G. Prota, G. Scherillo, and R. A. Nicolaus, Gazz. Chim. Ital. 98, 495 (1968).
4. E. Fattorusso, L. Minale, S. De Stefano, G. Cimino, and R. A. Nicolaus, Gazz. Chim. Ital. 99, 969 (1969).
5. These new compounds gave satisfactory elemental analyses.
6. R. Thomson, Angew. Chem. Internat. Edit. 13, 305 (1974).